Designing CO₂-responsive and degradable polymers for coating applications

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1. Introduction

Use of disposable plastic and coatings



- 9 billion tons of plastic manufactured globally in 2023
- ~50% of manufactured plastic is used for single-use purposes

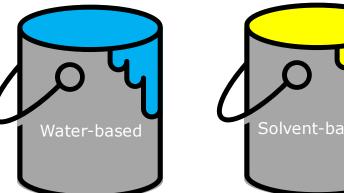


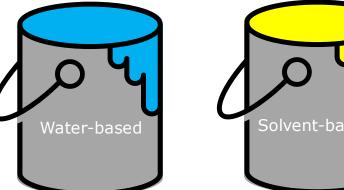
- Polymeric coatings have a significant presence in the commercial and industrial sectors
 - Global coatings market volume estimated to be 26.5 billion liters worth in 2009

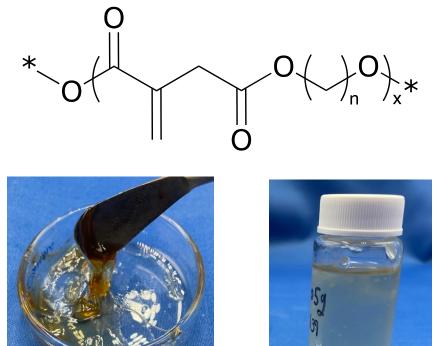
Related concerns of plastic and coating use





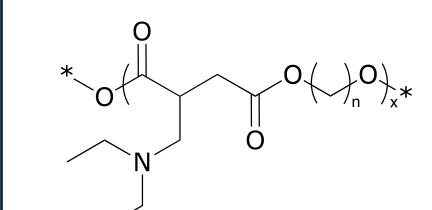






20.00 g scale

20.00 g scale



2.00 g scale

Reaction Scale	Mn	Mw	Polydispersity (Đ)
2.00 g	6409	8770	1.368
20.0 g	425	501	1.178

4. Results

Reaction M	In Mw	Polydispersity (Đ)
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- Persistent particles known as microplastics end up in the environment Microplastics cause harm to aquatic life and humans
- Coatings currently available are often lacking in performance (water-based) or use organic solvents as the bulk solvent (solvent-based)
 - Both coating types also result in persistent particles in the environment



2.00 g scale

2.00 g	2498	2723	1.090
20.0 g	523	638	1.221

Addressing the concerns related to plastics and coatings

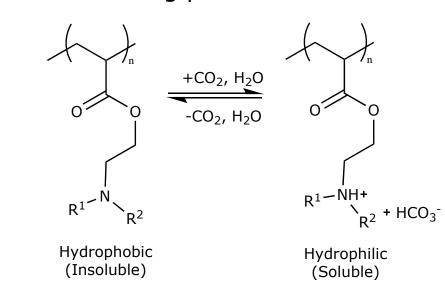
• The use of polymers that can degrade under specific conditions can offset microplastic contribution • Hydrolysis is an example of chemical degradation that can facilitate polymer breakdown

Degradation

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• The use of CO₂-responsive polymers allows carbonated water to be used as the bulk solvent

 Hydrophobic nature of the polymer in a neutral state offers superior coating performance



2. Goals of Research



Perform degradation and CO₂responsiveness tests for qualitative and quantitative

Tailor polymer properties for intended application



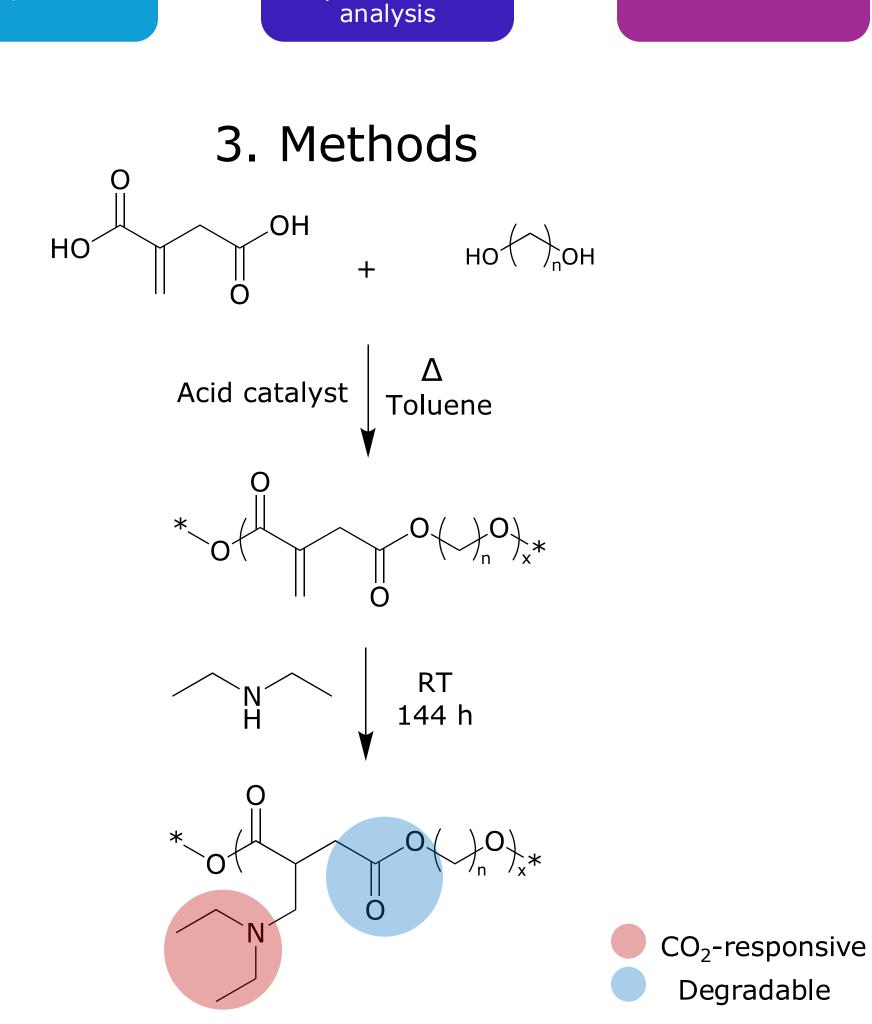
- 2 coats • 24 h drying time



 $+ CO_{2}$

• Droplet tests with DI water

- 24 h after droplet test
- Whitening can be seen on wood after applying water droplets to a wood surface



coated with polymer solution from 20.0 g scale batch

Molecular weight of the polymer is likely too low to achieve a hydrophobic coating upon surface application

5. Conclusions

- Condensation polymers bearing esters and tertiary amines can be synthesized using proposed method
- Scaling up the reaction results in a decrease in molecular weight up to ten-fold
- Coating performance was poor, but may be • improved by increasing molecular weight of the polymer

6. Future Work

- Adjust experimental set-up to achieve a higher molecular weight polymer
 - > Use a mechanical stirrer to improve reaction kinetics
 - Monitor reaction temperature throughout the reaction to ensure reflux is maintained
 - Monitor progress of polymerization via acid value

7. Acknowledgements

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- Dr. Michael Cunningham
- Dr. Tobias Robert
- Jessop Group members
- Cunningham Group members



Fraunhofer

8. References

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