

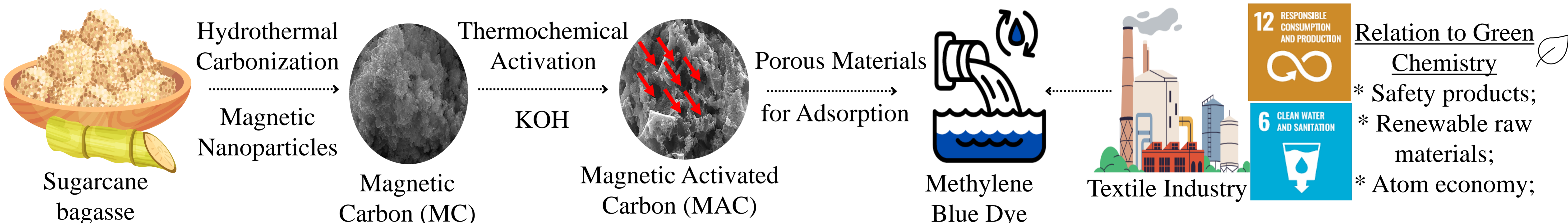
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INTRODUCTION

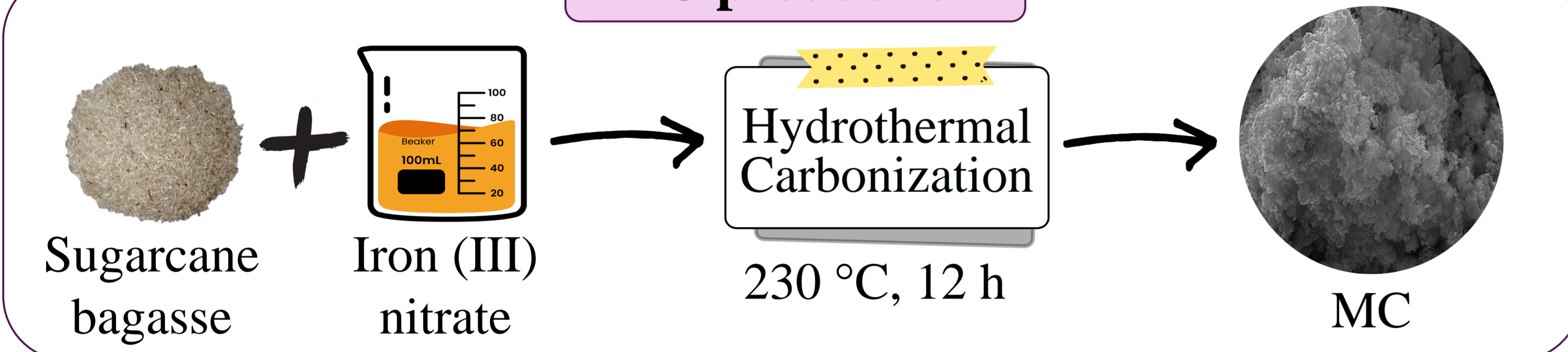


OBJECTIVE

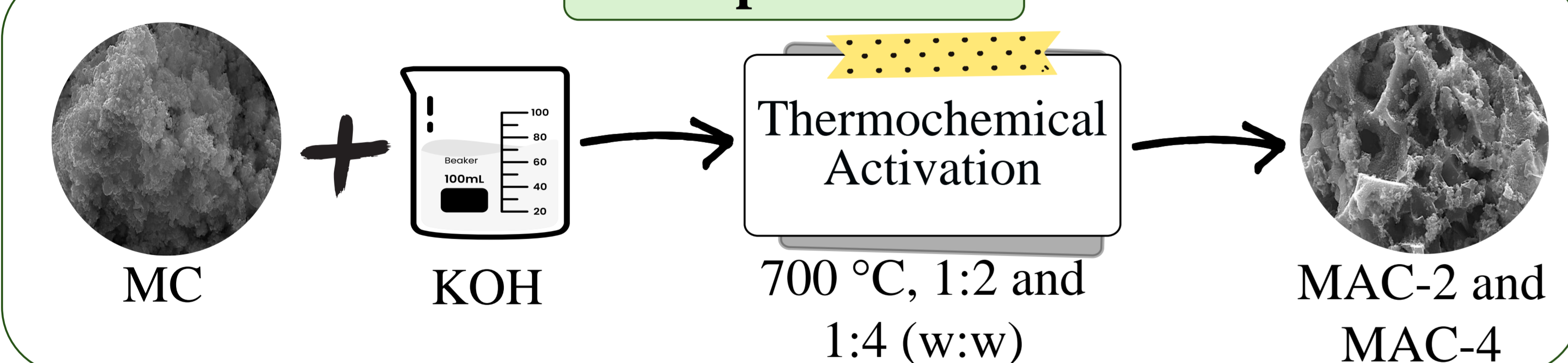
Evaluate the structure and texture of these KOH-activated materials and their performance in the adsorption of the methylene blue dye (MB).

MATERIALS AND METHODS

MC production



MAC production



Characterization

Structural and textural analysis

- **Functional groups:** Fourier Transform Infrared Spectroscopy (FTIR).
- **Surface area:** N₂ Adsorption-Desorption Isotherm (BET and BJH)

Application

Methylene Blue Adsorption

- 1.00 g L⁻¹ of MC and MACs in MB solution (300 mg L⁻¹);
- Stirring at 180 rpm for 4 h;
- MB determined by UV-Vis Spectrophotometry in $\lambda = 664$ nm.

RESULTS AND DISCUSSION

✓ Structural and textural analysis

- 3600 - 3125 cm⁻¹: O-H vibrations modes;
- 2290 - 2820 cm⁻¹: Symmetrical and asymmetric stretching of C-H bonds;
- 1735 cm⁻¹: C=O of the ketone and aldehyde groups;
- 1600 - 1511 cm⁻¹: related to the vibrations of C=C bonds;
- 1200 - 950 cm⁻¹: overlap of several bands but are mainly attributed to C-O-C vibrations of bonds;
- 550 cm⁻¹ and 440 cm⁻¹: stretching of Fe-O.
- Activation in the 1:2 ratio provided a greater specific surface area than the 1:4 ratio.

Table 1. Specific surface area (a_{BET}), pore volume (v_p) and adsorptive capacities (q_e) from MB.

Materials	a_{BET} (m ² g ⁻¹)	v_p (cm ³ g ⁻¹)	q_e (mg g ⁻¹)
MC	52.7	0.19	40.2
MAC-2	319.6	0.27	181.8
MAC-4	168.9	0.30	84.0



✓ Batch adsorption experiment

Activation with KOH in a ratio of 1:2 (MAC-2) showed significant adsorptive capacity of MB. Activation in a ratio 1:4 (MAC-4) was less effective.

CONCLUSION

Production of MAC with KOH in a 1:2 ratio was promising due to the presence of oxygenated functional groups, high specific surface area and adsorption capacity, while activation in a 1:4 ratio resulted in the destruction of some pores. The reuse of bagasse contributes to sustainable development and the circular bioeconomy.

REFERENCES

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ACKNOWLEDGMENTS



Process n° 264/2021

