



MANGANESE-CATALYZED NUCLEOPHILIC ADDITION OF ALDEHYDES TO CARBONYL COMPOUNDS VIA HYDRAZONE UMPOLUNG ON WATER

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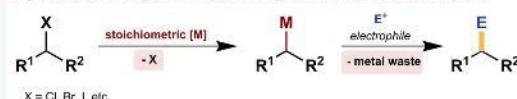
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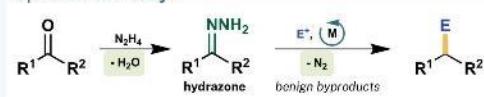


BACKGROUND

A | Traditional organometallic reagents in addition reactions



B | HOME-chemistry



Hydrazones as OrganoMetallic Equivalents (HOME-chemistry) still presents **two critical areas of improvement** to showcase its applicability as a greener alternative to classical methods:

- use non-precious and earth-abundant metals
- develop methods that are operationally simple



GREEN METRICS

Comparison of this work (Mn) **with other methods** for the nucleophilic addition of 1a to 2a via hydrazone umpolung

Benchmark	Ru ^a (ref. 1)	Fe ^b (ref. 2)	Ru ^c (ref. 3)	Mn ^d
earth-abundant metal	X	✓	X	✓
pre-synthesized catalyst	X	✓	X	X
solvent	THF	THF	H ₂ O	H ₂ O / neat
temperature (°C)	45	r.t.	70	50
atmosphere	N ₂	N ₂	Ar	air
additive	CsF	CsF	TBAC	none
yield (%)	94	99	71	84
atom economy (%)	88	88	88	88
reaction mass efficiency (%)	73	80	50	66
process mass efficiency ($\times 10^3$)*	2.8	2.6	4.4	3.4

^a[Ru(p-cymene)Cl₂]₂ (0.75 mol%) / dmpe (1.5 mol%)

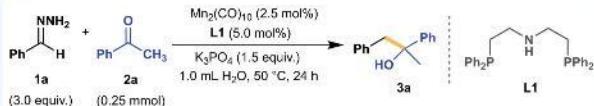
^b[Ru(p-cymene)Cl₂] (1.5 mol%) / PPh₃ (6.0 mol%)

*including purification

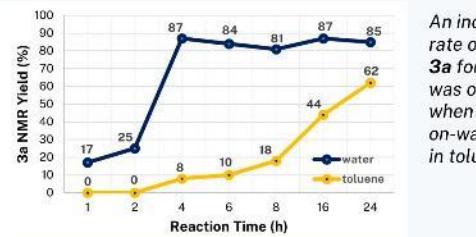
^bFe(dmpe)₂Cl₂ (5.0 mol%)

^cMn₂(CO)₁₀ (2.5 mol%) / Li (5.0 mol%)

REACTION DEVELOPMENT



Entry	Deviation	3a Yield (%)
1	no deviations	84
2	no pre-catalyst	0
3	no ligand	0
4	no pre-catalyst and ligand	0
5	open-air	80
6	1.25 equiv. 1a, 0.2 mL H ₂ O	84
7	3.0 equiv. 1a, neat	82

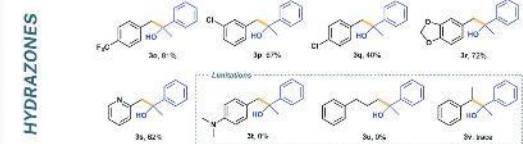
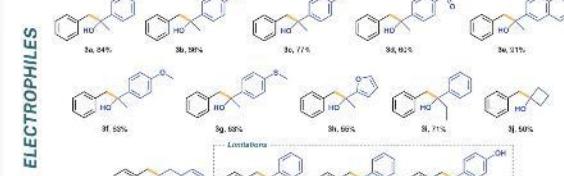


An increased rate of product 3a formation was observed when employing on-water versus in toluene.

Our method (Mn) works:

- even under open-air (**operationally simpler**)
- on water or neat (**safer solvent** or even **solventless**)
- even without the use of a phase-transfer catalyst
- with **commercially available** pre-catalyst and ligand

SUBSTRATE SCOPE



PROPOSED MECHANISM

