

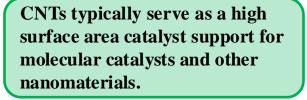
## Intrinsic catalytic activity of carbon nanotubes for electrocatalytic nitrate reduction to ammonia

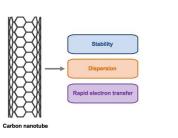
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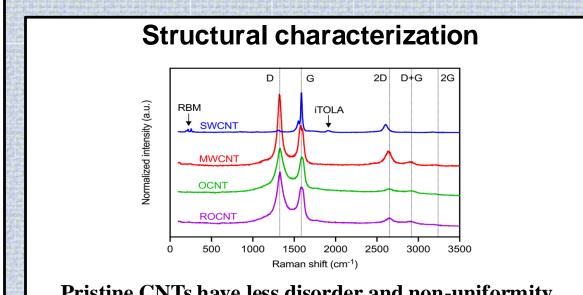
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## Introduction

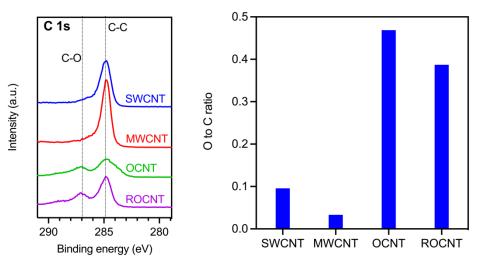
Nitrate  $(NO_3^-)$  ions are among the most widespread water pollutants impacting surface water and groundwater on a global scale. Electrochemical  $NO_3^-$  reduction is a promising approach to remove nitrate contaminants by converting them to valueadded fuels and chemical products, such as ammonia (NH<sub>3</sub>). There is growing interest in developing active and selective electrocatalysts for converting  $NO_3^-$  to  $NH_3$ . Carbon nanomaterials, including carbon nanotubes (CNTs), have been demonstrated to have a wide application in catalysis due to their desirable properties. This work investigates different types of CNTs as catalysts for the electroreduction of NO<sub>3</sub><sup>-</sup> to NH<sub>3</sub>.



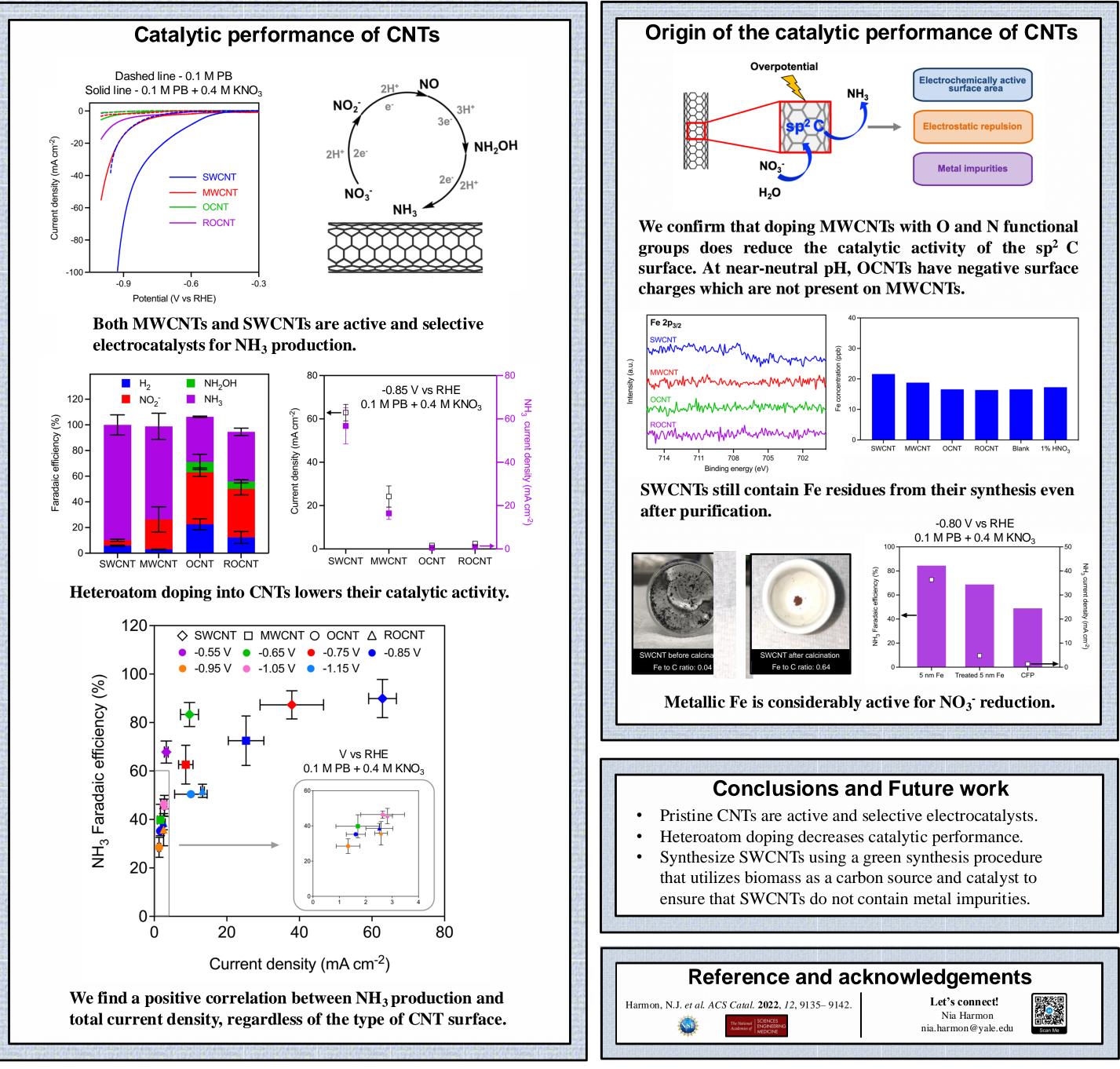








Partial restoration of the graphitic structure occurs when OCNTs are reduced to produce ROCNTs.



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