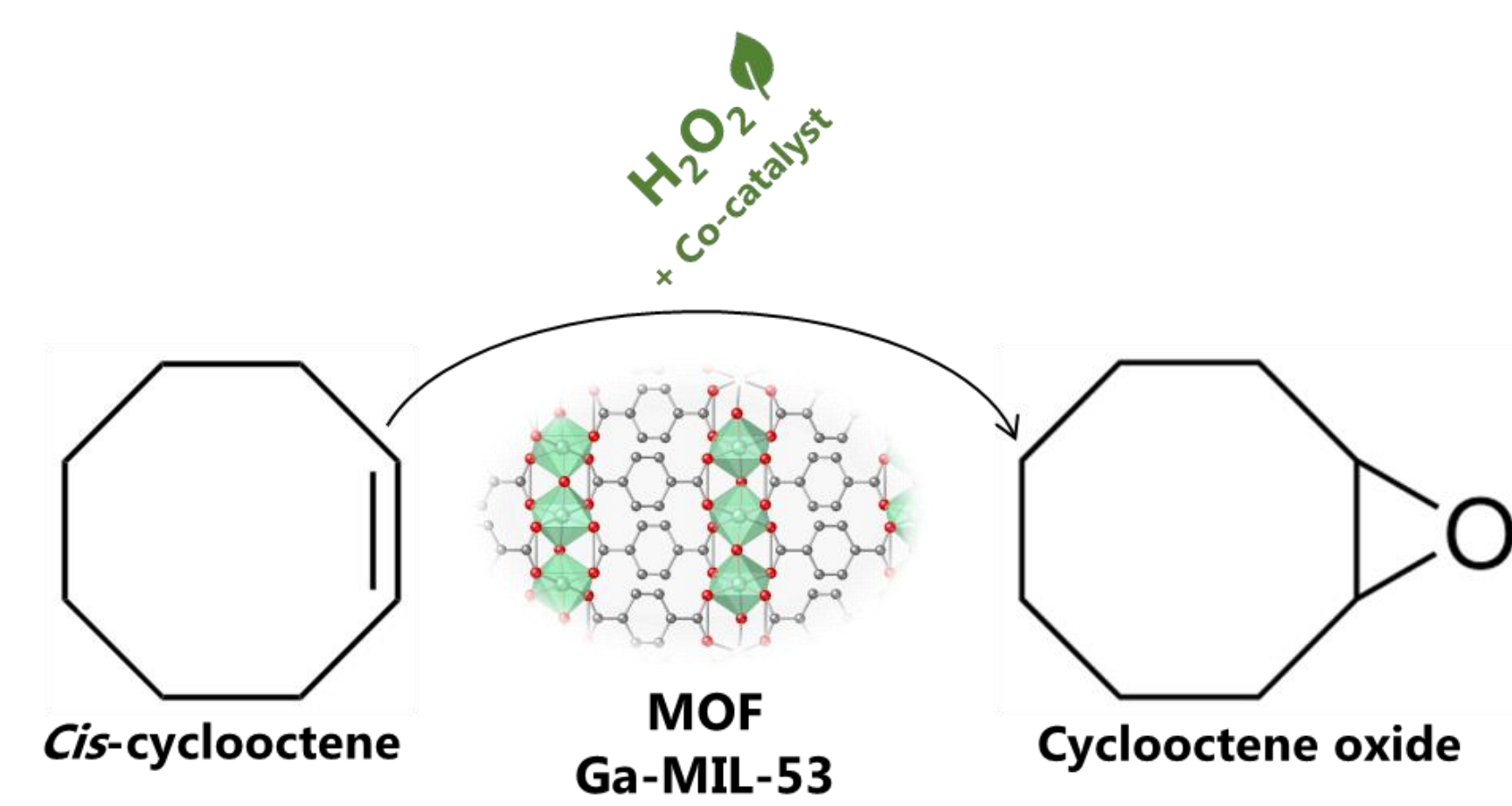


SYNTHESIS AND CHARACTERIZATION OF A ROBUST GA-MOF CATALYST FOR THE EPOXIDATION OF *CIS*-CYCLOOCTENE USING H₂O₂ AS A GREEN OXIDANT



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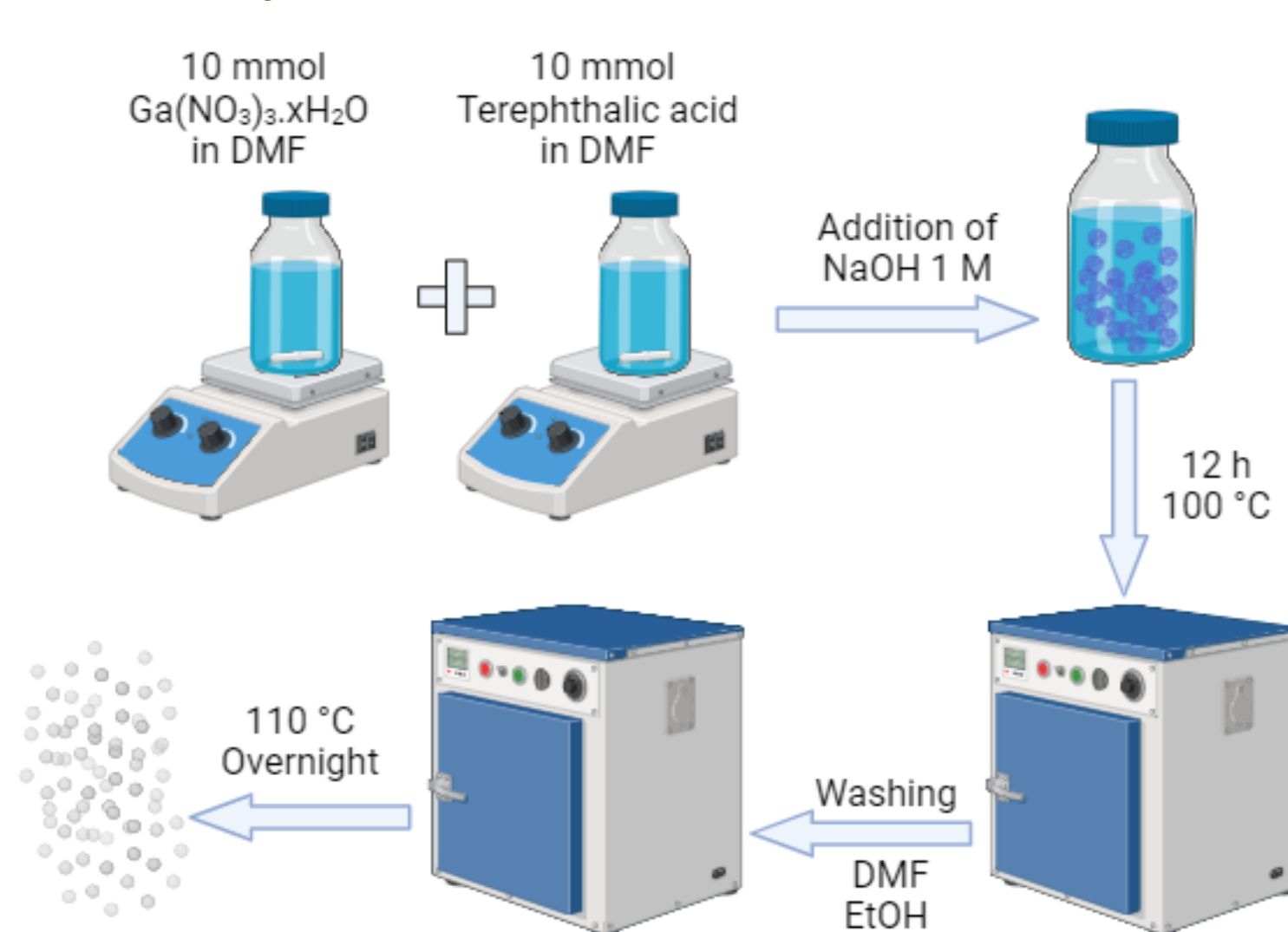
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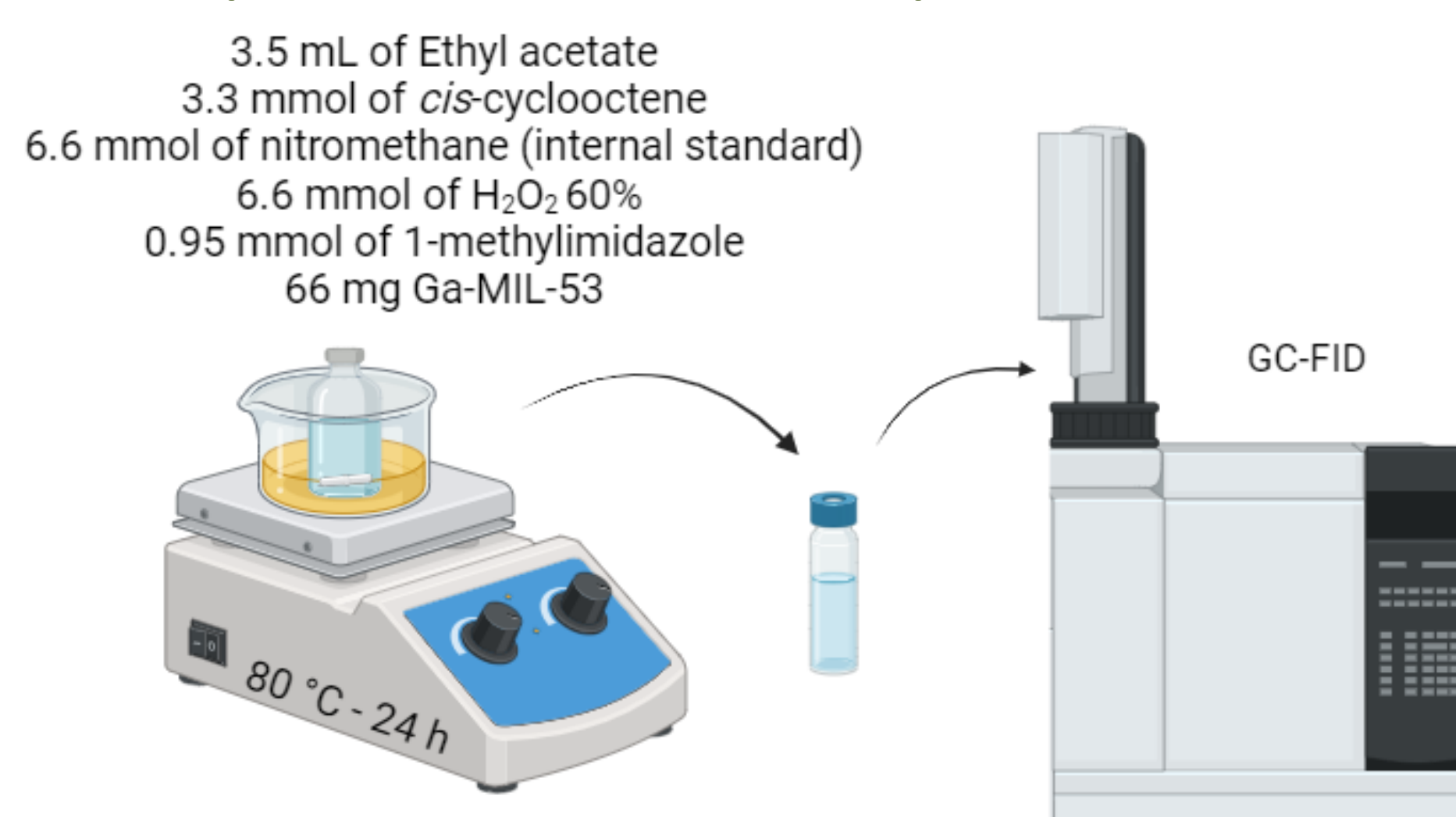
INTRODUCTION

An important route for producing epoxides, which are essential building blocks used in various industrial processes, is the catalytic oxidation of olefins ¹. Creating more sustainable and effective chemical procedures for these reactions is still a problem in modern chemistry. Because of their unique properties, using metal-organic frameworks (MOFs) as catalysts may present a viable alternative ².

Synthesis of Ga-MIL-53



Epoxidation of *cis*-cyclooctene



METHODS

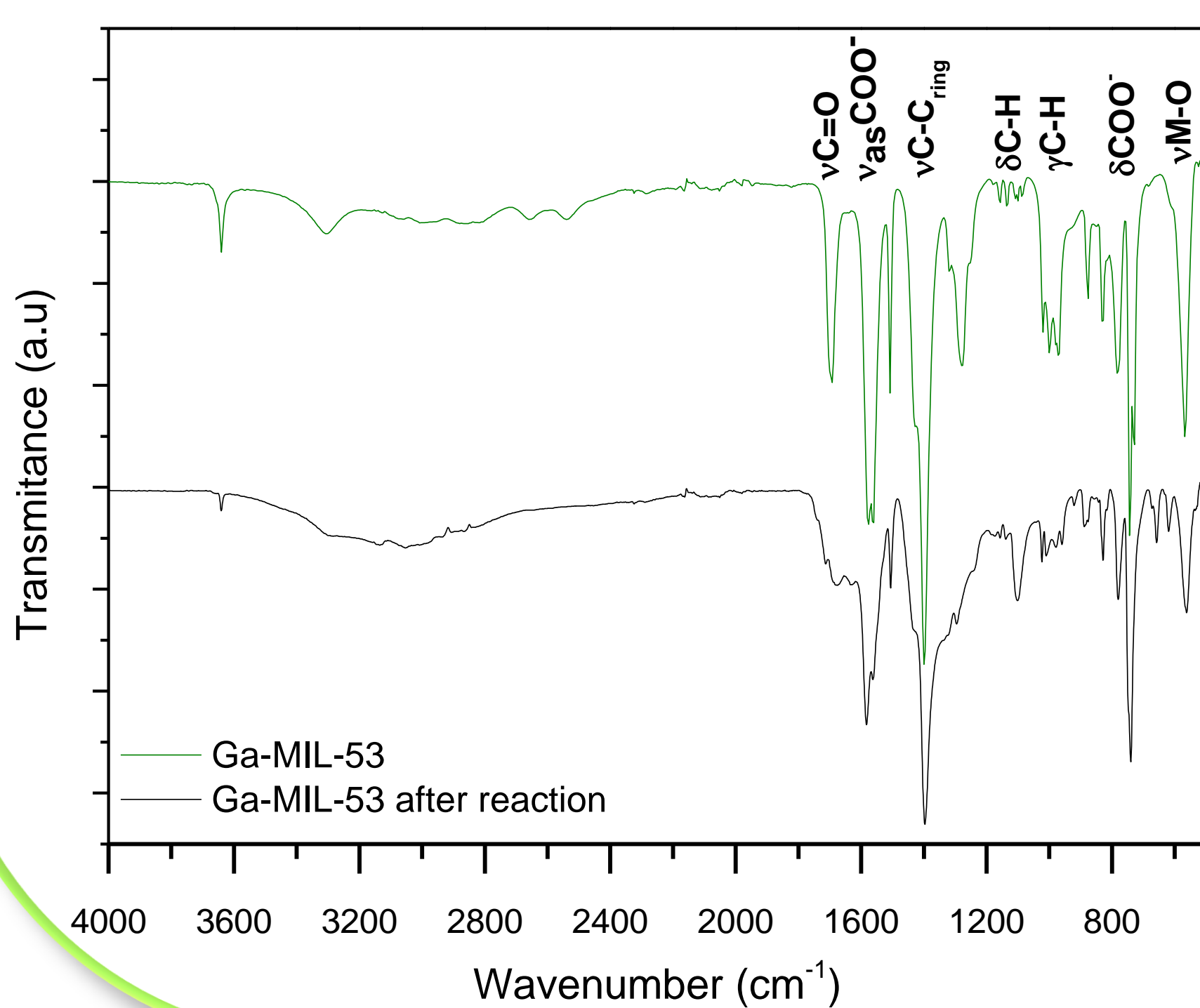
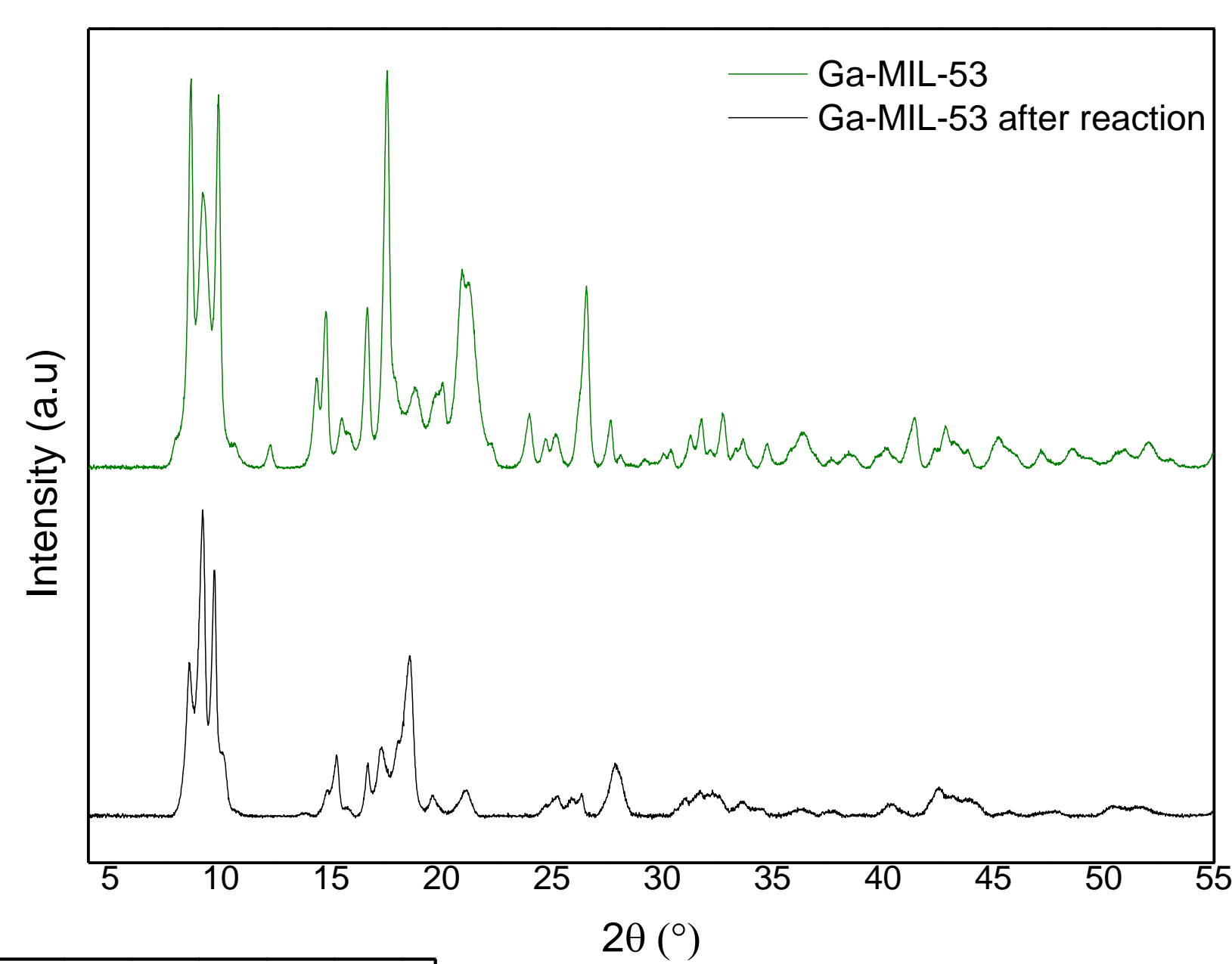
RESULTS

	C (%)	H (%)	N (%)	O (%)	Ga (%)
Chemical composition	38.6	2.4	-	33.9	25.1

Characterization of Ga-MOF

XRD

Figure 1 - XRD pattern for Ga-MOF before (red) and after reaction (blue).



FTIR

Figure 2 - FTIR spectra for Ga-MOF before (red) and after reaction (blue).

Catalytic performance – Epoxidation of *cis*-cyclooctene

Figure 3 – Kinetic profile of *cis*-cyclooctene epoxidation in 24 h of reaction. Reaction conditions: 66.6 mg of Ga-MIL-53, 0.95 mmol 1- methylimidazole, 3.5 mL of EtOAc, 3.3 mmol of *cis*-cyclooctene, 6.66 mmol of nitromethane, 6.66 mmol H₂O₂ 60%. 80 °C

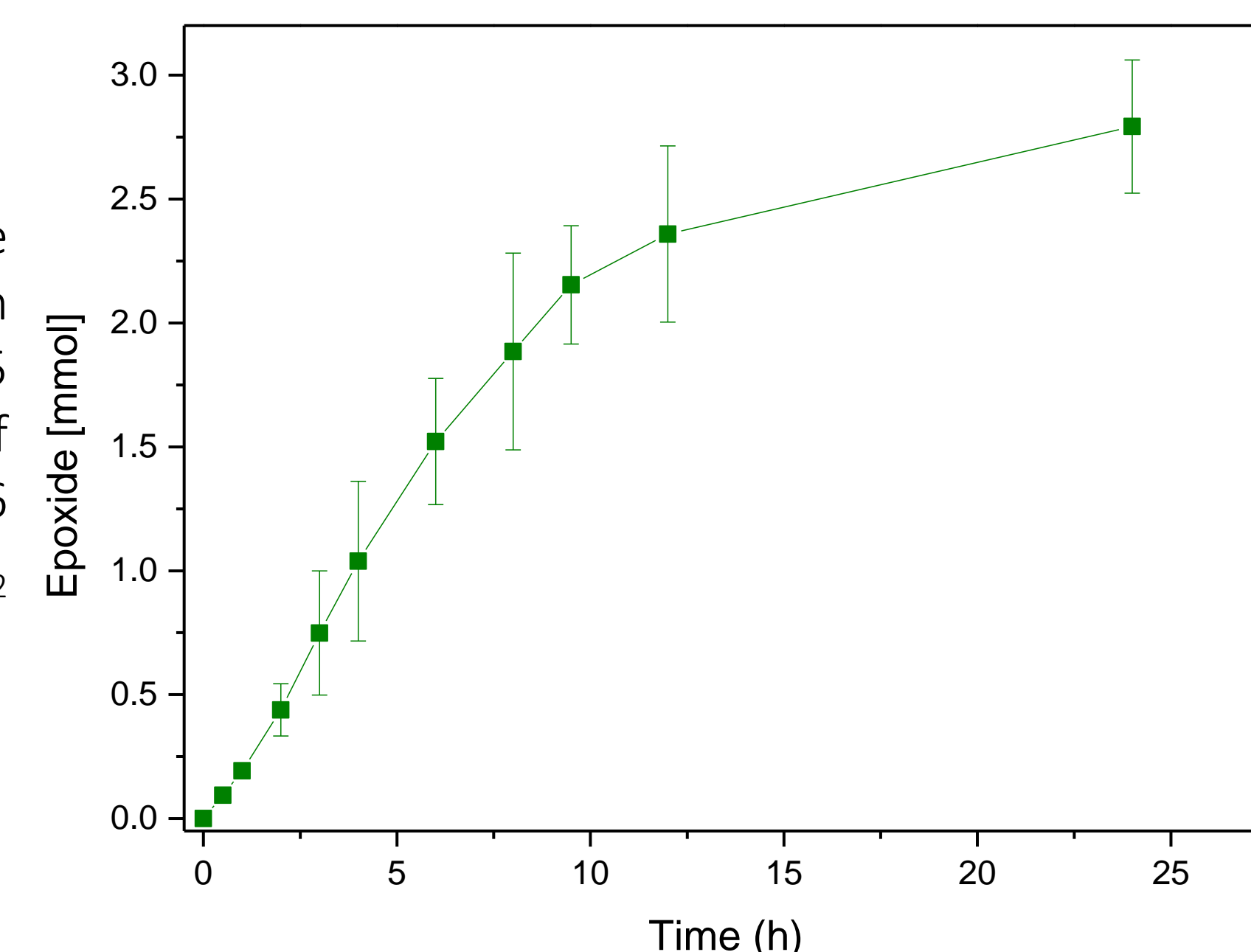


Table 1 – Conversion (%) and selectivity (%) of different *cis*-cyclooctene epoxidation reactions.

	Conversion (%)	Selectivity (%)
Blank	1.5	5.9
Ga-MIL-53	7.0	72.7
1-methylimidazole	16.9	52.3
Ga-MIL-53 + 1-methylimidazole	93.0	71.1
1° Cycle of reuse	81.6	93.0

○ Heterogenous nature confirmed by Sheldon's test ⁴ after 2 h of reaction.

CONCLUSION

The Ga-MIL-53 was successfully synthesized and evaluated for its catalytic performance in olefin epoxidation. The obtained material demonstrated significant efficiency in olefin epoxidation, which increased with a cooperative metal-ligand approach in the presence of 1-methylimidazole. Ga-MIL-53 produced 2.4 mmol of the corresponding epoxide, with a conversion rate of 93%, and epoxide selectivity of 71%, using H₂O₂ as the oxidant.

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